

Les ions polyatomiques

| Nom des ions | Formule et charge des ions |
|--|--|
| 1+ | |
| Ammonium | NH₄⁺ |
| 2+ | |
| Mercure | Hg ₂ ²⁺ |
| 1- | |
| Acétate | CH ₃ COO ⁻ |
| Benzoate | C ₆ H ₅ COO ⁻ |
| Chlorate | ClO ₃ ⁻ |
| Chlorite | ClO ₂ ⁻ |
| Cyanure | CN ⁻ |
| Dihydrogénophosphate | H ₂ PO ₄ ⁻ |
| Hydrogéocarbonate (bicarbonate) | HCO₃⁻ |
| Hydrogénooxalate | HOOC ⁻ COO ⁻ |
| Hydrogénosulfate (bisulfate) | HSO ₄ ⁻ |
| Hydrogénosulfite (bisulfite) | HSO ₃ ⁻ |
| Hydrogénésulfure (bisulfure) | HS ⁻ |
| Hydroxyde | OH⁻ |
| Hypochlorite | ClO ⁻ |
| Iodate | IO ₃ ⁻ |
| Nitrate | NO₃⁻ |
| Nitrite | NO ₂ ⁻ |
| Perchlorate | ClO ₄ ⁻ |
| Permanganate | MnO ₄ ⁻ |
| Stéarate | C ₁₇ H ₂₅ COO ⁻ |
| Thiocyanate | SCN ⁻ |
| 2- | |
| Carbonate | CO₃²⁻ |
| Chromate | CrO ₄ ²⁻ |
| Dichromate | Cr ₂ O ₇ ²⁻ |
| Hydrogénophosphate | HPO ₄ ²⁻ |
| Oxalate | C ₂ O ₄ ²⁻ |
| Peroxyde | O ₂ ²⁻ |
| Silicate | SiO ₃ ²⁻ |
| Sulfate | SO₄²⁻ |
| Sulfite | SO ₃ ²⁻ |
| Tartrate | C ₄ H ₄ O ₆ ²⁻ |
| Tétraborate | B ₄ O ₇ ²⁻ |
| Thiosulfate | S ₂ O ₃ ²⁻ |
| 3- | |
| Borate | BO ₃ ³⁻ |
| Phosphate | PO₄³⁻ |